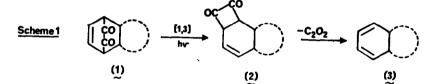
## PHOTO-BISDECARBONYLATION<sup>1</sup>. AN ENTRY INTO THE HIGHLY REACTIVE ISOBENZOSYSTEM: ISOINDENE

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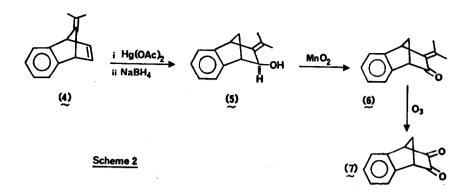
Rubin, Weiner and Scharf have recently reported<sup>2</sup> that alicyclic bridged  $\alpha$ -diketones (1) do not lose C<sub>2</sub>O<sub>2</sub> (or 2 x CO) directly upon irradiation at long wavelength (404, 436 nm), but undergo preliminary rearrangement to a cyclobutanedione (2), and this serves as the immediate precursor to the diene end-product (3) - Scheme 1.



In view of this finding it is of interest to compare the behavior of a system where the original  $\beta$ ,  $\gamma$ - olefin is locked into an aromatic system. In the event significant change does occur in the reaction pathway at longer wavelength (>300 nm), but short wavelength irradiation (254 nm) leads to efficient bisdecarbonylation. Thus this method has served as an entry into the isobenzo-series, illustrated herein by the generation of the highly reactive isoindene molecule (10) from  $\alpha$ -diketone (7).

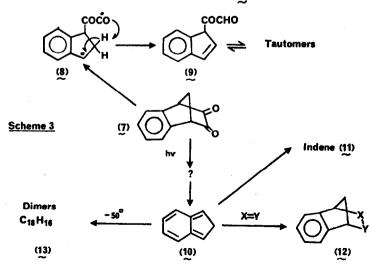
The benzonorbornenedione (7) was prepared via the sequence  $(4) \neq (7)^3$ . A significant step in this reaction is the regiospecific mercuration  $[Hg(OAc)_2/NaBH_4]^5$  of the  $\Delta^{2,3}$  olefin of the benzobicyclo [2.2.1] hepta-2,5-diene (4) which leads directly to the rearranged <u>exo</u>-enol (5). This reaction may well constitute the clearest example of the carbonium ion-type addition involving the Hg(II) reagent<sup>6,7</sup>, and contrasts with the simple norbornenes which do not rearrange under these conditions<sup>5</sup>. The oxidation step (5) to (6) can be achieved under Oppenauer conditions<sup>8</sup>, although we prefer activated manganese dioxide in benzene solution to generate the ketone (6). Other oxidizing agents cause deep-seated molecular reorganization<sup>9</sup>.

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In contrast with other  $\alpha$ -diketones<sup>1,2</sup>, long wavelength light (>300 nm; benzene or acetone, 0°C) did not effect smooth rearrangement to a product suitable for  $C_2O_2$  elimination, but yielded the ketoaldehyde (9)<sup>10</sup> as the major product (85%) together with indene (15%). The first step in this reaction is considered to involve Norish Type 1  $\alpha$ -cleavage to yield the diradical (8) which forms the ketoaldehyde (9) by hydrogen migration (arrows Scheme 3). A similar diradical could well be involved in the rearrangement of (1) to (2), since the finer mechanistic details of the rearrangement remain to be determined<sup>2</sup>. With the present example, formation of a compound related to (2) would involve dislocation of the aromatic ring and the alternative hydrogen migration step predominates.

Irradiation of the  $\alpha$ -diketone (7) with shorter wavelength u.v. light (250 nm) in acetone solution at 0°, reduced  $\alpha$ -cleavage to a minor role and raised the yield of indene to 70%. The ketoaldehyde (9) was shown not to be a source of indene under these conditions. Similar irradiations of (7) carried out in the presence of dienophile, resulted in the formation of  $[\pi^8_{\ 8} + \pi^2_{\ 8}]$ adducts of isoindene, together with the ketoaldehyde (9).



Thus <u>N</u>-methyl maleimide gave stereospecifically the <u>endo</u>-adduct of type  $\binom{12}{2}$ , m.p. 158° (p.m.r. in CDCl<sub>3</sub> :  $\delta$ 1.91, 2.11, dd J=8Hz, methylene;  $\delta$ 2.28, s, <u>N</u>-methyl;  $\delta$ 3.49, 3.81, m, methine protons, AA'XX' pattern typical of <u>endo</u>-adducts;  $\delta$ 7.19, m, aromatics), while dimethyl azodicarboxylate yielded a related adduct, an oil (p.m.r. in CDCl<sub>3</sub>:  $\delta$ 2.10, br, methylene,  $\delta$ 3.80, brs, methine,  $\delta$ 7.2-7.6, m, aromatics; the line broadening is due to the dynamic character of the <u>N</u>-substitutents of such adducts).

The above results are best explained by the intermediacy of isoindene<sup>11</sup> in the short wavelength photodecarbonylation reaction, which forms indene by a thermal 1,5-hydrogen shift in the absence of trapping agents and adducts in their presence.

Model experiments with 1,1,3-trideuteroindene ruled out indene to isoindene isomerisation as an alternative explanation. Thus <u>no</u> adducts were formed when this reagent was irradiated in the presence of <u>N</u>-methyl maleimide, and no deuterium incorporation was observed in the adducts formed when the diketone (7) was decomposed in the presence of both the trapping agent and the indene-d<sub>3</sub>.

While no direct evidence for isoindene could be observed (Pmr monitor) in low temperature (-50°) irradiations of the diketone (7); no indene was present either.

Work up of the complex mixture yielded two products ( $C_{18}H_{16}$ , m/e 232), which were isolated by P.L.C. Incorporation of indene- $\underline{d}_3$  in the low temperature experiment did not alter the course of the reaction and no deuterium was incorporated in the dimeric fraction. This suggests that 1,5-sigmatropic migration has been frozen at this temperature, thus allowing intermolecular dimerisation (photochemical or thermal) of the isoindene nucleus, rather than trapping of isoindene by partially isomerized indene<sup>12</sup>.

Experiments designed to utilize precursor (7) as a source of isoindene in matrix isolation studies are currently in progress, and will be reported, together with the structures of the above-mentioned dimers, in due course.

Acknowledgement: We thank the Australian Research Grants Committee for Financial Support.

## References and Footnotes

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- 11. Whether this is formed directly from diketone (7), or a rearranged isomer of type (2), has not been determined.
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